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~~What to read/Choose~~

~~your book/Book~~

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~~Lovers~~ Answers

Chemquest 33

Answers

Chemquest 33

Limiting Reactants

Answers the "limiting reactant" and oxygen is the excess

reactant. For each mole of C_3H_8 five moles of O_2 are required, so for 12.5 moles of C_3H_8 , the number of moles of O

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2 needed are
 $(12.5)(5) = 62.5$
moles.

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Limiting Reactants
Answers the "limiting
reactant" and oxygen
is the excess
reactant. For each

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mole of C_3H_8 five
moles of O_2 are
required, so for 12.5
moles of C_3H_8 , the
number of moles of O_2
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$(12.5)(5) = 62.5$
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Answers -

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Chemquest 33

Answers (Base the

answer to this

question on the

number of moles of

propane that actually

get combusted—which

is your answer to part

a.) 12 moles. For

every mole of

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propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$.
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Limiting Reactants

Answers the “limiting reactant” and oxygen is the excess

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$$(12.5)(5) = 62.5$$

moles.

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uyenyy.com

c) $3.1 \times 10^{-34} + 2.2 \times 10^{-33} =$ first, change exponents so both are either -34 or -33 ... sheet. Any substance listed on the periodic table. For example, phosphorus or nitrogen.

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Chemquest 33

Answer Key -

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ChemQuest 33 Name

T tlt'cym ,uTg.eTInT

F.uT'r \$\$ Date Hour

Information:

Determining if a Bond
is Polar In general the
greater the difference
in electronegativity
between t\i/o bonding

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atoms, the greater the polarity of the bond. A general rule of thumb is that if the difference in electronegativity is less than 0.5 then the bond is considered nonpolar.

,uTg.eTlnT tlt'cym

F.uT'r - LTHS

Answers

ChemQuest 33.

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ChemQuest 33.

Name: _____

_____.

Date:

_____.

Hour: _____.

Information: Limiting
Reactant. Again

consider the

combustion of

propane: $C_3H_8 + 5$

$O_2 \rightarrow 3 CO_2 + 4 H_2O$.

If you had 10 moles of
propane to burn, you

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would need 50 moles of oxygen according to the ratio in the balanced equation.

ChemQuest 33 -
Webs

ChemQuest 10

Name: _____ Date:

_____ Hour: _____

Information: Bohr's
Solar System Model
of the Atom All the

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planets are attracted to the sun by gravity. The reason that the Earth doesn't just float out into space is because it is constantly attracted to the sun. Similarly, the moon is attracted to the Earth by

Critical Thinking
Questions: Bohr's

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moles of O₂ are
required, so for 12.5

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33 moles of C_3H_8 , the
number of moles of O_2
needed are

$$(12.5)(5) = 62.5$$

moles. Since we have
more than 62.5 moles
(according to the
question we have

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Limiting Reactants

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33 Name T tl'cym

,uTg.eTlnT F.uT'r \$\$

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Answers

PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time. UNIT 12 - HW Practice

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Keys - ChemActivity

51: Cell Voltage -

ChemActivity 50:

Electrochemical Cell

UNIT 11 - HW

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Energy - ChemQuest

54: 2nd Law of

Thermodynamics

HW Keys - Roosevelt

High School AP

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Group, Inc. 8150

Corporate Park Drive

Suite 250 Cincinnati,

Ohio 45242 Phone:

+1 513 469 7555 Fax:

+1 513 469 7779

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info@chemquest.com

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JI ChemQuest 34

Name: Date Hour::F-F

Fluorine Gas Molar

mass = 38.0 g c.t- QI

Chlorine Gas Molar

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mass :: Br I Bromine
Liquid Molar mass : I-
I' Iodine Solid Molar
mass : Information:
Some Molecules and
Their States Critical
Thinking Questions
What type of force
exists between two Fz
molecules--{ispersion,
dipolar, or hydrogen
bonds?, ".eh"s#n
?(€l'5Feq-ece\$'1
What t5.pe of force

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33 Answers 3H8 five
moles of O₂ are
required, so for 12.5

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33 moles of C_3H_8 , the number of moles of O_2 needed are $(12.5)(5) = 62.5$ moles. Since we have more than 62.5 moles (according to the question we have

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Limiting Reactants

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33 Name T tl'tcym

,uTg.eTlnT F.uT'r \$\$

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ChemQuest 3 Name:

_____ Date: _____

Hour: _____

Information: Changes
in Matter Books are
made of matter. You
are made of matter.

“Matter” is a fancy

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word for the “stuff” of which all objects are made. Every day, matter is changed in different ways. For example, paper can be changed in many ways—it can be torn, folded, or burned. ...

Information : Changes
in Matter - Science! -
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33 Answer
A certain element exists as three different isotopes. 24.1% of all isotopes have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu.

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Chemistry
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Answers 24

ChemQuest 29 Monc
LcW s Structures tr l
Name Date: Hour:

Introduction

Questions Draw the
Lewis structures for
the following

molecules. Part c is
done for you. a) H_2O
(oxygen is central) b)
 SO_2 c) N_2 It t-t (a [Jrt
@ $N=N$ @ @ f\ @ 2

In question 1 above

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33 Answers
you should see three
different kinds
of bonds of bonds as
best you can.

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7b9ecd670